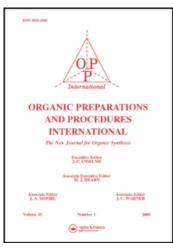
This article was downloaded by: On: *27 January 2011* Access details: *Access Details: Free Access* Publisher *Taylor & Francis* Informa Ltd Registered in England and Wales Registered Number: 1072954 Registered office: Mortimer House, 37-41 Mortimer Street, London W1T 3JH, UK



Organic Preparations and Procedures International Publication details, including instructions for authors and subscription information: http://www.informaworld.com/smpp/title~content=t902189982

1,3-DIPOLAR CYCLOADDITION OF AZIDOALKYLPHOSPHONATES AND CARBOXYLATES TO MALEIMIDE AND NAPHTHOQUINONE

F. Palacios^a; A. M. Ochoa de Retana^a; J. Pagalday^a ^a Departamento de Química Orgánica, Facultad de Farmacia, Universidad del País Vasco, Vitoria, SPAIN

To cite this Article Palacios, F., de Retana, A. M. Ochoa and Pagalday, J.(1995) '1,3-DIPOLAR CYCLOADDITION OF AZIDOALKYLPHOSPHONATES AND CARBOXYLATES TO MALEIMIDE AND NAPHTHOQUINONE', Organic Preparations and Procedures International, 27: 6, 625 – 635 To link to this Article: DOI: 10.1080/00304949509458520 URL: http://dx.doi.org/10.1080/00304949509458520

PLEASE SCROLL DOWN FOR ARTICLE

Full terms and conditions of use: http://www.informaworld.com/terms-and-conditions-of-access.pdf

This article may be used for research, teaching and private study purposes. Any substantial or systematic reproduction, re-distribution, re-selling, loan or sub-licensing, systematic supply or distribution in any form to anyone is expressly forbidden.

The publisher does not give any warranty express or implied or make any representation that the contents will be complete or accurate or up to date. The accuracy of any instructions, formulae and drug doses should be independently verified with primary sources. The publisher shall not be liable for any loss, actions, claims, proceedings, demand or costs or damages whatsoever or howsoever caused arising directly or indirectly in connection with or arising out of the use of this material.

1,3-DIPOLAR CYCLOADDITION OF AZIDOALKYLPHOSPHONATES AND CARBOXYLATES TO MALEIMIDE AND NAPHTHOQUINONE

F. Palacios*, A. M. Ochoa de Retana and J. Pagalday

Departamento de Química Orgánica, Facultad de Farmacia Universidad del País Vasco, Apartado 450, 01080 Vitoria, SPAIN

1-Aminoalkylphosphonate derivatives, isostere phosphonic analogues of α -aminoacids, are a new class of compounds with interesting biological properties¹ and thus are of significant interest in medicinal chemistry.¹⁻⁵ They are a unique class of simple mimetics of aminoacids and are used as antibacterial agents, neuromodulators, antibiotics, anticancer and antihypertensive drugs,¹ as well as neutral endopeptidase inhibitors², endotheling-converting enzyme inhibitors,³ irreversible inhibitors for blood coagulation,⁴ and as other enzyme inhibitors.⁵ Furthermore, heterocycles derived from maleimides are utilized in the synthesis of polymers,⁶ while quinone derivatives exhibit high anticancer activity.⁷ Maleimide and naphthoquinone are widely used as dipolarophiles in 1,3-dipolar cycloaddition reactions.⁸ In spite of the potential interest of the cycloadducts, there is little information about the reaction of both these dipolarophiles with azides.⁹ To the best of our knowledge, the reaction with maleimide is restricted to the use of aryl,¹⁰ silylazides¹¹ and azidoalkylindoles,¹² while in the case of quinones the reaction with aryl¹³ and glicosylazides¹⁴ has been reported.

In recent years, we have been involved in the chemistry of phosphazenes¹⁵ derived from azides¹⁶ and enamino functionalized phosphorus derivatives,¹⁷ in their use in the synthesis of the hete-rocyclic¹⁸ and acyclic¹⁹ compounds as well as in the 1,3-cycloaddition reaction of azides derived from aminophosphonates with alkynes and electron-rich olefines such as enamines.²⁰ We now describe the intermolecular cycloaddition of azidoalkyl-phosphonates and carboxylates with electron-poor double bonds such as maleimide and naphtoquinone.

Maleimide 1 adds azidoalkylphosphonates 2 in tetrahydrofuran at room temperature (48 hrs) to give isolable triazolinealkylphosphonates 3 ($E = PO(OEt)_2$), members of the 2,3,4,7-tetraazabicyclo[3.3.0]octane family. When $R \neq H$ a mixture of two diastereoisomers is obtained. The spectral data (Table 1) are in agreement with bicyclic compounds 3. The vicinal coupling constants are in the range of 10.9-11.2 Hz for H-1 and H-5 protons, evidence for the *syn* configuration of the ring junction protons¹¹ and they are consistent with a stereospecific *cis* addition²¹ of maleimide to the azide. Thermolysis of triazolines 3 triggers nitrogen evolution and leads to the enamines 4. Spectroscopic data are consistent with the proposed structure, showing in the ¹H NMR spectrum of 4a a singlet at δ 5.10 ppm

^{© 1995} by Organic Preparations and Procedures Inc.

for the heterocyclic proton at the 4-position. Azides derived from aminoesters such as azidoalkylcarboxylate **5** react similarly with maleimide yielding triazolinealkylcarboxylate **6** ($E = CO_2Et$). Thermal nitrogen elimination from triazoline gives the corresponding aminosubstituted maleimide **7** (Scheme 1).

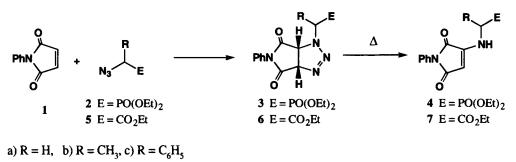
Cmpd	³¹ P NI	MR ¹ H NMR (δ , ppm)	¹³ C NMR (δ , ppm)
	(δ, pp		
3a	19.6	1.35 (t, 6H, ${}^{3}J_{HH} = 7.1$ Hz, CH ₃), 4.19 (m, 5H, OCH ₂ , CHN), 4.57 (dd, 1H, ${}^{2}J_{PH} = 16.3$ Hz, ${}^{2}J_{HH} = 13.1$ Hz, CHN), 4.89 (d, 1H, ${}^{3}J_{HH} = 10.9$ Hz, CH), 5.67 (d, 1H, ${}^{3}J_{HH} = 10.9$ Hz, CH), 7.22-7.48 (m, 5H, arom.)	16.4 (CH ₃), 43.8 (d, ${}^{1}J_{PC} = 155.1$ Hz, CH ₂ N), 57.5 (CH), 62.9 (OCH ₂), 81.9 (CH), 126.2-130.8 (C arom.), 169.4 (C=O), 170.9 (C=O)
3b	23.5	1.32 (t, 6H, ${}^{3}J_{HH} = 7.1$ Hz, CH ₃), 1.72 (dd, 3H, ${}^{3}J_{PH} = 16.3$ Hz, ${}^{3}J_{HH} = 7.5$ Hz, CH ₃), 4.15 (m, 4H, OCH ₂), 4.69 (m, 1H, CHN), 4.91 (d, 1H, ${}^{3}J_{HH} = 10.9$ Hz, CH), 5.74 (d, 1H, ${}^{3}J_{HH} = 10.9$ Hz, CH), 7.25-7.46 (m, 5H, arom.)	15.1 (CH ₃), 16.4 (CH ₃), 52.1 (d, ${}^{1}J_{PC} = 149.1$ Hz, CHN), 57.2 (CH), 62.8 (OCH ₂), 82.6 (CH), 126.2-130.8 (C arom.), 168.8 (C=O), 171.3 (C=O)
3b'	21.8	1.35 (m, 6H, CH ₃), 1.79 (dd, 3H, ${}^{3}J_{PH} = 16.1 \text{ Hz}, {}^{3}J_{HH} = 7.3 \text{ Hz}, \text{CH}_{3}$), 4.16 (m, 4H, OCH ₂), 4.30 (m, 1H, CHN), 4.96 (d, 1H, ${}^{3}J_{HH} = 11.2 \text{ Hz}$, CH), 5.70 (d, 1H, ${}^{3}J_{HH} = 11.2 \text{ Hz}$, CH), 7.24-7.47 (m, 5H, arom.)	15.0 (CH ₃), 16.4 (CH ₃), 52.3 (d, ${}^{1}J_{PC} = 152.9$ Hz, CHN), 59.0 (CH), 62.8 (OCH ₂), 81.3 (CH), 126.2-130.9 (C arom.), 169.5 (C=O), 171.6 (C=O)
3c	17.5	1.18 (t, 3H, ${}^{3}J_{HH} = 7.1$ Hz, CH ₃), 1.25 (t, 3H, ${}^{3}J_{HH} = 7.1$ Hz, CH ₃), 4.12 (m, 4H, OCH ₂), 4.28 (d, 1H, ${}^{3}J_{HH} = 10.9$ Hz, CH), 5.35 (d, 1H, ${}^{2}J_{PH} = 24.2$ Hz, CHN), 5.61 (d, 1H, ${}^{3}J_{HH} = 10.9$ Hz, CH), 7.30-7.48 (m, 10H, arom.)	16.2 (CH ₃), 56.9 (CH), 60.5 (d, ${}^{1}J_{pC}$ = 152.6 Hz, CHN), 63.5 (OCH ₂), 80.7 (CH), 126.2-132.3 (C arom.), 168.9 (C=O), 170.6 (C=O)
3c'	17.6	1.25 (m, 6H, CH ₃), 4.10 (m, 4H, OCH ₂), 4.76 (d, 1H, ${}^{3}J_{HH} = 10.9$ Hz, CH), 5.29 (d, 1H, ${}^{2}J_{PH} = 21.9$ Hz, CHN), 5.73 (d, 1H, ${}^{3}J_{HH} = 10.9$ Hz, CH), 7.35-7.55 (m, 10H, arom.)	16.2 (CH ₃), 59.3 (CH), 61.7 (d, ${}^{1}J_{PC} = 152.1$ Hz, CHN), 63.5 (OCH ₂), 81.0 (CH), 126.0-132.9 (C arom.), 169.8 (C=O), 170.7 (C=O)
4a	20.5	1.37 (m, 6H, CH ₃), 3.56 (dd, 2H, ² J _{<i>p</i>H} = 12.8 Hz, ³ J _{<i>HH</i>} =6.2 Hz, CH ₂ N), 4.21 (m, 4H, OCH ₂), 5.10 (s, 1H, CH=), 5.74 (broad t, 1H, ³ J _{<i>HH</i>} =6.2 Hz, NH) ^a , 7.26-7.47 (m, 5H, arom.)	16.5 (CH ₃), 40.5 (d, ${}^{1}J_{PC} = 158.7$ Hz, CH ₂ N), 63.0 (OCH ₂), 86.4 (CH=), 125.8-131.8 (C arom.), 148.9 (C=), 165.9 (C=O), 170.8 (C=O)

Table 1. Spectral Data of Triazolines (3, 6) and Enamines (4, 7).

1,3-DIPOLAR CYCLOADDITION OF AZIDOALKYLPHOSPHONATES AND CARBOXYLATES

Table 1. Continued					
Cmpd	³¹ Ρ NMF (δ, ppm)		¹³ C NMR (δ, ppm)		
4 b	23.5	1.30 (m, 6H, CH ₃), 1.47 (dd, 3H, ${}^{3}J_{PH} = 16.6 \text{ Hz}, {}^{3}J_{HH} = 7.3 \text{ Hz}, \text{CH}_{3}$), 3.60 (m, 1H, CHN), 4.13 (m, 4H, OCH ₂), 5.02 (s, 1H, CH=), 5.54 (d, 1H, ${}^{3}J_{HH} = 8.3 \text{ Hz}, \text{NH}^{a}$, 7.24-7.41 (m, 5H, arom.)	15.1 (CH ₃), 16.5 (CH ₃), 47.5 (d, ${}^{1}J_{PC} = 160.1$ Hz, CHN), 62.9 (OCH ₂), 86.2 (CH=), 125.7-131.6 (C arom.), 147.7 (C=), 165.9 (C=O), 170.7 (C=O)		
4c		1.15 (t, 3H, ${}^{3}J_{HH} = 7.1$ Hz, CH ₃), 1.31 (t, 3H, ${}^{3}J_{HH} = 7.1$ Hz, CH ₃), 3.75-4.14 (m, 4H, OCH ₂), 4.62 (dd, 1H, ${}^{2}J_{PH} = 21.4$ Hz, ${}^{3}J_{HH} = 7.3$ Hz, CHN), 4.86 (s, 1H, CH=), 6.28 (t, 1H, ${}^{3}J_{HH} = 7.3$ Hz, NH) ^a , 7.27-7.45 (m, 10H, arom.)	16.4 (CH ₃), 56.7 (d, ¹ J _{<i>PC</i>} = 153.3 Hz, CHN), 63.9 (OCH ₂), 88.4 (CH=), 125.8-132.8 (C arom.), 147.6 (C=), 166.0 (C=O), 170.5 (C=O)		
6a		1.29 (t, 3H, ${}^{3}J_{HH} = 7.1$ Hz, CH ₃), 4.23 (q, 2H, ${}^{3}J_{HH} = 7.1$ Hz, OCH ₂), 4.60 (d, 1H, ${}^{3}J_{HH} = 10.9$ Hz, CH), 4.62 (d, 1H, ${}^{2}J_{HH} = 18.3$ Hz, CHN), 4.83 (d, 1H, ${}^{2}J_{HH} = 18.3$ Hz, CHN), 5.79 (d, 1H, ${}^{3}J_{HH} = 10.9$ Hz, CH), 7.25-7.48 (m, 5H, arom.)	14.1 (CH ₃), 49.0 (CH ₂ N), 57.4 (CH), 61.9 (OCH ₂), 82.6 (CH), 126.3-129.3 (C arom.), 168.5 (COO), 169.3 (C=O), 170.9 (C=O)		
7a		1.33 (t, 3H, ${}^{3}J_{HH} = 7.1$ Hz, CH ₃), 3.97 (d, 2H, ${}^{3}J_{HH} = 5.4$ Hz, CH ₂ N), 4.30 (q, 2H, ${}^{3}J_{HH} = 7.1$ Hz, OCH ₂), 5.01 (s, 1H, CH=), 5.97 (broad t, 1H, ${}^{3}J_{HH} = 5.4$ Hz, NH) ^a , 7.29-7.48 (m, 5H, arom.)	14.1 (CH ₃), 45.3 (CH ₂ N), 62.2 (OCH ₂), 86.5 (CH=), 125.8-131.6 (C arom.), 148.0 (C=), 165.7 (C=O), 167.9 (COO), 170.7 (C=O)		

^a Exchangeable proton with D₂O





Cmpnd	³¹ P N		¹³ C NMR (δ, ppm)
10a	(δ, pp		141 (CH) 542 (CH N) (10
104		1.30 (t, 3H, ${}^{3}J_{HH} = 7.1$ Hz, CH ₃), 4.26 (q, 2H, ${}^{3}J_{HH} = 7.1$ Hz, OCH ₂), 4.56 (s, 2H, CH ₂ N), 5.94 (d, 1H, ${}^{3}J_{HH} = 10.9$ Hz, CH), 6.75 (d, 1H, ${}^{3}J_{HH} = 10.9$ Hz, CH), 7.72-8.46 (m, 4H, arom.)	14.1 (CH ₃), 54.3 (CH ₂ N), 61.9 (OCH ₂), 112.3 (CH), 122.3-137.5 (C arom., CH), 167.3 (C=O), 167.8 (COO), 187.2 (C=O)
11a	14.5	1.28 (t, 6H, ${}^{3}J_{HH} = 7.1$ Hz, CH ₃), 4.16 (m, 4H, OCH ₂), 5.30 (d, 2H, ${}^{2}J_{PH} = 13.5$ Hz, CH ₂ N), 7.78-8.30 (m, 4H, arom.)	16.2 (CH ₃), 45.4 (d, ${}^{1}J_{PC} = 152.1$ Hz, CH ₂ N), 63.7 (OCH ₂), 127.3-145.1 (C arom., =C, C=), 175.1 (C=O), 176.4 (C=O)
11b	18.3	1.27 (t, 3H, ${}^{3}J_{HH} = 7.1$ Hz, CH ₃), 1.34 (t, 3H, ${}^{3}J_{HH} = 7.1$ Hz, CH ₃), 2.04 (dd, 3H, ${}^{2}J_{PH} = 15.8$ Hz, ${}^{3}J_{HH} = 7.4$ Hz, CH ₃), 4.17 (m, 4H, OCH ₂), 5.96 (m, 1H, CHN), 7.80-8.36 (m, 4H, arom.)	15.2 (CH ₃), 16.4 (CH ₃), 53.3 (d, ¹ J _{PC} = 154.1 Hz, CHN), 63.7 (OCH ₂), 127.4-145.0 (C arom., =C, C=), 175.5 (C=O), 176.7 (C=O)
11c	14.7	1.21 (t, 3H, ${}^{3}J_{HH} = 7.1$ Hz, CH ₃), 1.26 (t, 3H, ${}^{3}J_{HH} = 7.1$ Hz, CH ₃), 4.08 (q, 2H, ${}^{3}J_{HH} = 7.1$ Hz, OCH ₂), 4.23 (t, 2H, ${}^{3}J_{HH} = 7.1$ Hz, OCH ₂), 6.94 (d, 1H, ${}^{2}J_{PH} = 21.7$ Hz, CHN), 7.35-8.28 (m, 9H, arom.)	16.3 (CH ₃), 61.33 (d, ${}^{1}J_{PC} = 155.1$ Hz CHN), 64.2 (OCH ₂), 127.3-145.0 (C arom., =C, C=), 175.4 (C=O), 176.5 (C=O)
12a		1.31 (t, 3H, ${}^{3}J_{HH} = 7.1$ Hz, CH ₃), 4.29 (q, 2H, ${}^{3}J_{HH} = 7.1$ Hz, OCH ₂), 5.59 (s, 2H, CH ₂ N), 7.77-8.36 (m, 4H, arom.)	14.1 (CH ₃), 51.0 (CH ₂ N), 62.8 (OCH ₂), 127.3-145.2 (C arom., =C, C=), 165.2 (COO), 175.3 (C=O), 176.5 (C=O)
13a	18.9	1.32 (t, 6H, ${}^{3}J_{HH} = 7.1$ Hz, CH ₃), 3.73 (dd, 2H, ${}^{2}J_{PH} = 12.4$ Hz, ${}^{3}J_{HH} = 6.4$ Hz, CH ₂ N), 4.16 (m, 4H, OCH ₂), 7.56-7.74 (m, 5H, arom., CH=), 9.15 (m, 1H, NH) ^a	16.4 (CH ₃), 45.4 (d, ${}^{1}J_{PC} = 155.4$ Hz, CH ₂ N), 63.3 (OCH ₂), 121.5-133.6 (C arom.), 139.8 (C=), 152.2 (CH=), 189.6 (C=O), 193.5 (C=O)
13b	21.9	1.35 (t, 6H, ${}^{3}J_{HH} = 7.1$ Hz, CH ₃), 1.61 (dd, 3H, ${}^{2}J_{PH} = 16.7$ Hz, ${}^{3}J_{HH} = 7.3$ Hz, CH ₃), 3.73 (m, 1H, CHN), 4.19 (m, 4H, OCH ₂), 7.60-7.78 (m, 5H, arom., CH=), 9.20 (m, 1H, NH) ^a	15.2 (CH ₃), 16.5 (CH ₃), 52.2 (d, ¹ J _{PC} = 157.1 Hz, CHN), 63.4 (OCH ₂), 121.4-133.5 (C arom.), 139.7 (C=), 150.5 (CH=), 189.8 (C=O), 193.6 (C=O)
l3c	18.1	1.14 (t, 3H, ${}^{3}J_{HH} = 7.1$ Hz, CH ₃), 1.21 (t, 3H, ${}^{3}J_{HH} = 7.1$ Hz, CH ₃), 3.84-4.10 (m, 4H, OCH ₂), 4.74 (dd, 1H, ${}^{2}J_{PH} = 19.4$ Hz, ${}^{3}J_{HH} = 8.2$ Hz, CHN), 7.32-7.72 (m, 10H, arom., CH=), 9.80 (m, 1H, NH) ^a	16.2 (CH ₃), 60.9 (d, ${}^{1}J_{PC} =$ 153.6 Hz, CHN), 63.8 (OCH ₂), 121.3-133.5 (C arom.), 139.7 (C=), 151.1 (CH=), 189.6 (C=O), 193.6 (C=O)

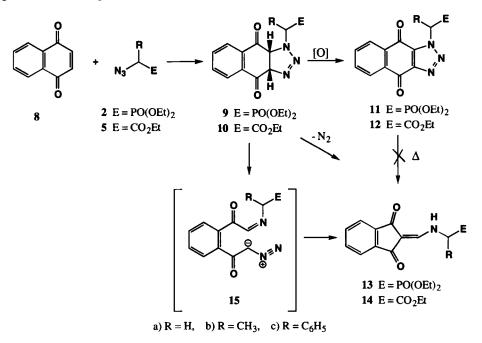
Table 2. Spectral Data of Triazoles (11, 12) and Enaminones (13, 14).

1,3-DIPOLAR CYCLOADDITION OF AZIDOALKYLPHOSPHONATES AND CARBOXYLATES

Table 2. Continued							
Cmpnd	³¹ P NMR	¹ H NMR (δ, ppm)	¹³ C NMR (δ, ppm)				
	(ð, ppm)						
1 4 a	1.36	$\overline{\mathbf{b}}$ (t, 3H, ${}^{3}\mathbf{J}_{HH} = 7.1$ Hz, CH ₃),	4.0 (CH ₃), 50.1 (CH ₂ N),				
	4.20	$(d, 2H, {}^{3}J_{HH} = 6.1 \text{ Hz}, CH_{2}N),$	162.1 (OCH ₂), 121.3-133.3 (C arom.), 139.7 (C=), 152.0 (CH=), 167.8				
	4.32	$2 (q, 2H, {}^{3}J_{HH} = 7.1 \text{ Hz}, OCH_{2}),$					
		2-7.76 (m, 5H, arom., CH=),	(COO), 189.8 (C=O),				
	9.29	(m, 1H, NH) ^a	193.2 (C=O)				

a) Exchangeable proton with D₂O.

The cycloaddition of azides derived from α -aminophosphonates 2 and aminoacids 5 to 1,4naphthoquinone gives, depending on the reaction conditions, triazoles 11, 12 and indan-1,3-diones 13, 14 (Scheme 2). Thus, when azides 2 or 5 and naphthoquinone are allowed to react at room temperature in tetrahydrofuran for two weeks the 1-phosphanoalkyl- 11 or 1-ethoxycarboxyalkyl-naphto(2.3d)triazol-4,9-dione 12 are isolated in low yields. However, the reaction of phosphorus containing azide 2 with 1,4-naphtoquinone in refluxing THF allows the isolation of triazole derivative 11 (75%); while when the reaction is run in refluxing toluene a (80/20) mixture of triazole 11 and the enamine 13, and traces of triazoline 9 is obtained (Scheme 2). It is noteworthy that triazole compounds 11 were thermally stable and do not undergo nitrogen elimination even after prolonged boiling in refluxing toluene (72 hrs). The treatment of azidoalkylcarboxylate 5 with quinone 8 (room temperature, refluxing THF, or toluene) gives similar results to that observed in the case of azide 2.



Scheme 2

These results suggest that the process is initiated by the 1,3-cycloaddition reaction giving rise to the bicyclic triazolines 9 or 10, which subsequently undergo either dehydrogenation to yield substituted triazoles 11 or 12. Elimination of a nitrogen molecule affords the contracted ring compounds 13 or 14. Formation of these enamine derivatives is favored by the use of a higher reaction temperature. Thermolysis of triazolines to give indan-1,3-diones has been previously reported^{14,22} and could be explained through 1,3-dipolar cycloreversion²³ of bicyclic triazolines 9 or 10 to give a new dipole 15. The elimination of N₂ of the diazoimine 15, followed by ring closure could undergo the ring contraction from the six-membered quinoinic ring of triazoline 9 to the indan-1,3-diones 13 in a fashion similar to that suggested for the thermolysis of bicyclic triazolines.^{24,25}

In conclusion, new 1,3-dipolar cycloaddition reactions of azides derived from α -aminophosphonates and α -aminoacids with maleimides and quinones are described and new families of ethoxycarboxyland phosphoryl-alkyltriazolines 3 and 6, triazoles 11 and 12 and enamines 4, 7, 13, 14 are reported.

EXPERIMENTAL SECTION

Melting points were determined with a Buchi SPM-20 apparatus and are uncorrected. Analytical TLC was performed on 0.25 mm silica gel plates (Merck). Visualization was accomplished by UV light and iodine. Solvents for extraction and chromatography were technical grade and distilled from the indicated drying agents. All solvents used in reactions were freshly distilled from appropriate drying agents before use; toluene (Na); tetrahydrofuran (Na). All other reagents were recrystallized or distilled as necessary. Column (flash) chromatography was carried out on silica gel (Merck, 70-230 mesh). Mass spectra were obtained on a Hewlett Packard 5890 spectrometer. ¹H and ¹³C NMR experiments were recorded on a Bruker 75 MHz spectrometer using tetramethylsilane (0.00 ppm) or chloroform (7.26 ppm) as an internal reference in CDCl₃ solutions. ³¹P NMR spectra were recorded at 120 MHz with 85% phosphoric acid as an external reference. Chemical shifts are given in ppm (δ); multiplicities are indicated by s (singlet), d (doublet), dd (double-doublet), t (triplet) q (quadruplet) or m (multiplet). Coupling constants, *J*, are reported in Hertz. Mass spectra (EI) were obtained with a ionization voltage of 70 eV. Data are reported in the form *m/z* (intensity relative to base = 100).

General Procedure for the Cycloaddition of N-Phenylmaleimide with Azidoalkyl-phosphonates and carboxylates.- To a solution of diethyl 1-azidoalkylphosphonate²⁰ 2 (R = H, CH₃, Ph) or ethyl 1azidocarboxylate 5 (R=H) (3 mmol) in THF (15 mL) was added dropwise with stirring a solution of N-phenylmaleimide 1 (3 mmol, 0.52 g), and the reaction mixture was stirred at room temperature for 48 hrs. The solvent was evaporated and the residue was purified by flash column chromatography (silica gel) with ether:hexane as eluent and recrystallized to give triazolines 3 or 6.

2,N-[1-(Diethoxyphosphonylmethyl)]-7,N-Phenyl-2,3,4,7-tetraazabicyclo[3.3.0]octane (3a).-Reaction with diethyl 1-azidomethylphosphonate 2a (3 mmol, 0.58 g) gave: 0.93 g (85%) of the compound 3a as a yellow solid, mp.79-80°. Recrystallized from hexane: CH_2Cl_2 ; MS, *m/z*: 338 (M⁺-N₂, 25%).

Anal. Calcd. for C₁₅H₁₉N₄O₅P: C, 49.17; H, 5.23; N, 15.30. Found: C, 49.10; H, 5.29; N, 15.33.

2,*N*-[**1**-(**1**-Diethoxyphosphonylethyl)]-**7**,*N*-Phenyl-**2**,**3**,**4**,**7**-tetraazabicyclo[**3**.**3**.**0**]octane (**3**b).-Reaction with diethyl 1-azido-1-ethylphosphonate **2b** (3 mmol, 0.62 g) gave: 1.03 g (90%) of the mixture of two diastereoisomers **3b** and **3b'** isolated by flash column chromatography (silica gel) with ether:hexane as eluent. Data for **3b**: Obtained as a white solid, mp. 98-99°. Recrystallized from hexane:CH₂Cl₂; MS, m/z: 352 (M⁺-N₂, 8%).

Anal. Calcd. for $C_{16}H_{21}N_4O_5P$: C, 50.51; H, 5.57; N, 14.74. Found: C, 50.93; H, 5.49; N, 14.79 Data for **3b'**: Obtained as a white solid, mp. 99-100°. Recrystallized from hexane:CH₂Cl₂; MS, *m/z*: 352 (M⁺-N₂, 12%).

Anal. Calcd. for C16H21NaO5P: C, 50.51; H, 5.57; N, 14.74. Found: C, 50.60; H, 5.63; N, 14.68

2,N-[1-(Diethoxyphosphonyl-1-phenylmethyl)]-7,N-Phenyl-2,3,4,7-tetraazabicyclo-[3.3.0]octane (3c).- Reaction with diethyl 1-azido-1-phenylmethylphosphonate 2c (3 mmol, 0.81 g) gave: 1.06 g (80%) of the mixture of two diastereoisomers 3c and 3c' isolated by flash column chromatography (silica gel) with ether:hexane as eluent. Data for 3c: Obtained as a yellow solid, mp. 128-129°. Recrystallized from hexane:CH₂Cl₂; MS, m/z: 414 (M⁺-N₂, 9%).

Anal. Calcd. for $C_{21}H_{23}N_4O_5P$: C, 56.99; H, 5.24; N, 12.67. Found: C, 57.06; H, 5.18; N,12.72 Data for **3c**': Obtained as a yellow solid, mp. 129-130°. Recrystallized from hexane: CH_2Cl_2 ; MS, *m/z*: 414 (M⁺-N₂, 2%).

Anal. Calcd. for C₂₁H₂₃N₄O₅P: C, 56.99; H, 5.24; N, 12.67. Found: C, 57.15; H, 5.33; N, 12.58

2,*N*-[**1**-(Ethoxycarbonylmethyl)]-7,*N*-Phenyl-2,3,4,7-tetraazabicyclo[3.3.0]octane(6a).- Reaction with ethyl 1-azidomethylcarboxylate **5a** (3 mmol, 0.39 g) gave: 0.81 g (90%) of the compound **6a** as a yellow solid, mp. 141-142°. Recrystallized from hexane: CH_2Cl_2 ; MS, m/z: 274 (M⁺-N₂, 80%).

Anal. Calcd. for C₁₄H₁₄N₄O₄: C, 55.61; H, 4.67; N, 18.54. Found: C, 55.50; H, 4.73; N, 18.59

General Procedure for the Thermolysis of Triazolinealkyl-phosphonates and carboxylates.- A solution of triazolinealkyl-phosphonate 3 or -carboxylate 6 (3 mmol) in toluene (15 mL) was stirred in refluxing for 15 hrs. The solvent was evaporated and the residue was purified by flash column chromatography (silica gel) with ether:hexane as eluent to give enamines 4 or 7.

3-(Diethoxyphosphonylmethylamine)-*N*-Phenylmaleimide (4a).- Thermolysis of 2,*N*-[1-(diethoxyphosphonylmethyl)]-7,*N*-phenyl-2,3,4,7-tetrabicyclo[3.3.0]octane **3a** (3 mmol, 1.10 g) gave: 0.91 g (90%) of the compound **4a** as a yellow solid, mp. 124-125°. Recrystallized from hexane:CH₂Cl₂; MS, m/z: 338 (M⁺, 62%).

Anal. Calcd. for C₁₅H₁₉N₂O₅P: C, 53.25; H, 5.66; N, 8.28. Found: C, 53.21; H, 5.71; N, 8.36

3-(1-Diethoxyphosphonylethylamine)-N-Phenylmaleimide (4b).- Thermolysis of 2,N-[1-(1-diethoxyphosphonylethyl)]-7,N-phenyl-2,3,4,7-tetrabicyclo[3.3.0]octane **3b** (3 mmol,1.14 g) gave: 0.92 g (87%) of the compound **4b** as a yellow solid, mp. 116-117°. Recrystallized from hexane:CH₂Cl₂; MS, *m/z*: 352 (M⁺, 37%).

Anal. Calcd. for C₁₆H₂₁N₂O₅P: C, 54.53; H, 6.01; N, 7.95. Found: C,54.60; H, 6.10; N, 7.89

3-(Diethoxyphosphonyl-1-phenylmethylamine)-N-Phenylmaleimide (4c).- Thermolysis of 2,*N*-[1-(diethoxyphosphonyl-1-phenylmethyl)]-7,*N*-phenyl-2,3,4,7-tetrabicyclo[3.3.0]octane **3c** (3 mmol, 1.33 g) gave:1.03 g (83%) of the compound **4c** as a yellow solid, mp.133-134°. Recrystallized from hexane:CH₂Cl₂; MS, *m/z*: 414 (M⁺, 13%).

Anal. Calcd. for C₂₁H₂₃N₂O₅P: C, 60.85; H, 5.60; N, 6.76. Found: C,60.80; H, 5.67; N, 6.82

3-(Ethoxycarbonylmethylamine)-*N***-Phenylmaleimide (7a)**.- Thermolysis of 2,*N*-[1-(ethoxycarbonylmethyl)]-7,*N*-phenyl-2,3,4,7-tetrabicyclo[3.3.0]octane **6a** (3 mmol, 0.91 g) gave:0.70 g (85%) of the compound **7a** as a yellow solid, mp.119-120°. Recrystallized from hexane: CH_2Cl_2 ; MS, *m/z*: 274 (M⁺, 87%).

Anal. Calcd. for C₁₄H₁₄N₂O₄: C, 61.29; H, 5.15; N, 10.22. Found: C, 61.38; H, 5.08; N, 10.29

General Procedure for the Preparation of Triazolealkyl-phosphonates and carboxylates.- To a solution of diethyl 1-azidoalkylphosphonate 2 (R=H, CH_3 , Ph) or ethyl 1-azidoalkylcarboxylate 5 (R=H) (3 mmol) in THF (15 mL) was added dropwise with stirring a solution of 1,4-naphthoquinone 8 (3 mmol, 0.47 g), and the reaction mixture was stirred in refluxing THF for 20 hrs. The solvent was evaporated and the residue was purified by flash column chromatography (silica gel) with ether:hexane as eluent to give triazoles 11 or 12.

1-(Diethoxyphosphonylmethyl)naphtho[2.3-d]triazol-4,9-dione (11a).- Reaction with diethyl 1azidomethylphosphonate 2a (3 mmol, 0.58 g) gave:0.79 g (75%) of the compound 11a as a brown solid, mp.145-146°. Recrystallized from hexane:CH₂Cl₂; MS, m/z: 349 (M⁺, 27%).

Anal. Calcd. for C₁₅H₁₆N₃O₅P: C, 51.56; H, 4.62; N, 12.03. Found: C, 51.47; H, 4.54; N, 12.12

1-(1-Diethoxyphosphonylethyl)naphtho[2.3-d]triazol-4,9-dione (11b).- Reaction with diethyl 1azido-1-ethylphosphonate 2b (3 mmol, 0.62 g) gave: 0.76 g (70%) of the compound 11b as a brown solid, mp.114-115°. Recrystallized from hexane:CH₂Cl₂; MS, *m/z*: 363 (M⁺, 10%).

Anal. Calcd. for C₁₆H₁₈N₃O₅P: C, 52.88; H, 5.00; N, 11.57. Found: C, 52.90; H,4.95; N, 11.50

1-(Diethoxyphosphonyl-1-phenylmethyl)naphtho[2.3-d]triazol-4,9-dione (11c).- Reaction with diethyl 1-azido-1-phenylmethylphosphonate 2c (3 mmol, 0.81 g) gave: 0.97g (76%) of the compound 11c as a brown solid, mp.168-169°. Recrystallized from hexane:CH₂Cl₂; MS, *m*/*z*: 425 (M+, 5%).

Anal. Calcd. for C₂₁H₂₀N₃O₅P: C, 59.28; H, 4.74; N, 9.88. Found: C, 59.38; H, 4.70; N, 9.81

1-(Ethoxycarbonylmethyl)naphtho[2.3-d]triazol-4,9-dione (12a).- Reaction with ethyl 1azidomethylcarboxylate 5a (3 mmol, 0.39 g) gave: 0.68 g (80%) of the compound 12a as a brown solid, mp.154-155°. Recrystallized from hexane: CH_2Cl_3 ; MS, m/z: 285 (M⁺, 2%).

Anal. Calcd. for C₁₄H₁₁N₃O₄: C, 58.93; H, 3.89; N, 14.74. Found: C, 58.99; H, 3.81; N, 14.65

General Procedure for the Cycloaddition of Naphthoquinone with azidoalkyl-phosphonates and -carboxylates in refluxing toluene.- To a solution of diethyl 1-azidoalkylphosphonate 2 (R=H, CH₃, Ph) (3 mmol) in toluene (15 mL) was added dropwise with stirring a solution of 1,4-naphthoquinone 8 (3 mmol, 0.47 g), and the reaction mixture was stirred in refluxing for 36 hrs. Concentration in vacuum gave the mixture of triazolealkyl-phosphonates 11, indan-1,3-diones 13 (80/20), isolated by flash column chromatography (silica gel) with ether:hexane as eluent, and traces of triazoline compounds 9. When ethyl 1-azidoalkylcarboxylate 5 (R=H) is used a (75/20/5) mixture of triazole 12, indan-1,3-dione 14 and triazoline 10 are obtained.

Products 11, and 12, their physical constant and spectral data are identical with those previously obtained.

2-(Diethoxyphosphonylmethylaminomethylen)-indan-1,3-dione (13a).- Reaction with diethyl 1azidomethylphosphonate **2a** (3 mmol, 0.58 g) gave a (80/20) mixture of **11a** and **13a** (70%). Data of **13a**: 0.14 g, obtained as a brown solid, mp.120-121°. Recrystallized from hexane:CH₂Cl₂; MS, m/z: 323 (M⁺, 35%).

Anal. Calcd. for C15H18NO5P: C, 55.71; H, 5.62; N, 4.33. Found: C, 55,67; H, 5.69; N, 4.27

2-(1-Diethoxyphosphonylethylaminomethylen)-indan-1,3-dione (13b).- Reaction with diethyl 1azido-1-ethylphosphonate 2b (3 mmol, 0.62 g) gave a (78/22) mixture of 11b and 13b (75%). Data of 13b: 0.15 g, obtained as a syrup. Rf (ethyl acetate)= 0.19; MS, *m/z*: 337 (M⁺, 50%).

Anal. Calcd. for C₁₆H₂₀NO₅P: C, 56.96; H, 5.98; N, 4.15. Found: C, 56.85; H, 5.92; N, 4.21

2-(Diethoxyphosphonyl-1-phenylmethylaminomethylen)-indan-1,3-dione (13c).- Reaction with diethyl 1-azido-1-phenylmethylphosphonate **2c** (3 mmol, 0.81 g) gave a (81/19) mixture of **11c** and **13c** (70%). Data of **13c**: 0.17 g, obtained as a syrup. Rf (ethyl acetate:hexane=1:1)= 0.19; MS, m/z: 399(M⁺, 6%).

Anal. Calcd. for C21H22NO5P: C, 63.14; H, 5.56; N, 3.51. Found: C, 63.22; H, 5.47; N, 3.59

1-(Ethoxycarbonylmethyl)naphtho[2.3-d]triazoline-4,9-dione (10a) and 2-(Ethoxycarbonylmethylaminomethylen)-indan-1,3-dione (14a).- Reaction with ethyl 1-azidomethylcarboxylate 5a (3 mmol, 0.39 g) gave a (75/20/5) mixture of 12a, 14a , and 10a (82%). Data of 10a: 0.04 g, obtained as a brown solid, mp. 84-85°. Recrystallized from hexane: CH_2Cl_2 ; MS, m/z: 259 (M⁺-N₂, 11%). *Anal.* Calcd. for $C_{14}H_{13}N_3O_4$: C, 58.53; H, 4.56; N, 14.63. Found: C, 58.61; H, 4.49; N, 14.70. Data of 14a: 0.12 g, obtained as a syrup. Rf (ethyl acetate:hexane=1:1)= 0.33; MS, m/z: 259 (M⁺, 10%). *Anal.* Calcd. for $C_{14}H_{13}NO_4$: C, 64.84; H, 5.06; N, 5.41. Found: C, 64.91; H, 5.00; N, 5.60

Acknowledgements.- J. P. thanks the Consejería de Educación y Universidades del Gobierrno Vasco for a predoctoral fellowship. Financial support by Dirección General de Investigación Científica y Técnica (DGICYT, PB 93-0501) and by the Consejería de Educación y Universidades del Gobierno Vasco (PGV 94-36) is gratefully acknowledged.

REFERENCES

- For reviews see: A. D. F. Toy and E. N. Walsh in "Phosphorus Chemistry in Everyday Living", American Chemical Society, Washington DC, 1987, p. 333. Dhawar and D. Redmore, "Phosphorus & Sulfur", 32, 119 (1987); R. E. Hoagland in "Biologically Active Natural Products" ed. by H. G. Culter; ACS Symposium Series 380, American Chemical Society, Washington, DC, 1988, p. 182; P. Kafarski and B. Lejezak, "Phosphorus & Sulfur", 63, 193 (1991).
- S. De Lombaert, M. D. Erion, J. Tan, L. Blanchard, L. El-Chehabi, R. D. Ghai, Y. Sakane, C. Berry and J. A. Trapari, J. Med. Chem., 37, 498 (1994); S. De Lombaert and M. D. Erion, U. S. Patent 5.294.632. Chem. Abstr., 121, 281220 (1994).
- 3. K. Ishikawa, T. Fukami, T. Hayama, K. Matsuyama, K. Noguchi and M. Yano. Jpn. Kokai Tokkyo Koho JP 05.148.277, Chem. Abstr, 120, 7634 (1994).
- 4. J. Oleksyszyn, B. Boduszek, C. M. Kam and J. C. Powers, J. Med. Chem., 37, 226 (1994).

- R. A. Nugent, M. Murphy, S. T. Schalachter, C. J. Dunn, R. J. Smith, N. D. Staite, L. A. Galinet, S. K. Shields, D. G. Aspar, K. A. Richard and N. A. Rohloff, *ibid.*, 36, 134 (1993); R. B. Baudy, L. P. Greenblatt, I. L. Jirkovsky, M. Conklin, R. J. Russo, D. R. BramLett, T. A. Emrey, J. T. Simmonds, D. M. Kowal, R. P. Stein and R. P. Tasse, *ibid.*, 36, 331 (1993).
- 6. Y. Gilliams and G. Smets, Makromol. Chem., 117, 1 (1968); ibid , 128, 263 (1969).
- J. S. Driscoll, G. F. Hazard, H. B. Wood and A. Goldin, *Cancer Chemother. Rep.*, part 2, 4 (2), 1, (1974), 1. H. D. H. Showalter, D. W. Fry, W. R. Leopold, J. W. Lown, J. A. Plambeck and K. Reszka, *Anti-Cancer Drug Design*, 1, 73 (1986).
- For excellent reviews see: a) "1,3-Dipolar Cycloaddition Chemistry". Ed. A. Padwa. J. Willey, New York. 1984. b) A. Padwa in "Comprehensive Organic Synthesis" ed. B. M. Trost, I. Fleming, Pergamon Press, 1991, vol. 4, ed. M. F. Semmelhack, p. 1069.
- 9. For a review see W. Lwowski, in ref. 8a, Vol. 1, p. 559.
- S. J. Davis and C. S. Rondestvedt, Chem. Ind. (London), 845 (1956); R. Huisgen, G. Szeimies and L. Möbius, Chem. Ber., 100, 2494 (1967).
- 11. S. S. Washburne, W. R. Peterson and D. A. Berman, J. Org. Chem., 37, 1738 (1972).
- Y. Tamura, M. W. Chun, K. Ohno, S. Kwon and M. Ikeda, *Chem. Pharm. Bull. Jpn*, 26, 2874 (1978).
- 13. L. Wolff and R. Hercher, Ann., **399**, 274 (1913); F. D. Chattaway and G. D. Parke, J. Chem. Soc., **127**, 1387 (1925); L. F. Fieser and J. L. Hartwell, J. Am. Chem. Soc., **57**, 1479 (1935).
- G. Alonso, M. Fuertes, M. T. García-López, F. G. de las Heras, J. M. Infante and M. Stud, Eur. J. Med. Chem., 13, 155 (1978).
- 15. For a review see J. Barluenga and F. Palacios, Org. Prep. Proced. Int., 23, 1 (1991).
- F. Palacios, D. Aparicio, J. M. de los Santos, I. Pérez de Heredia and G. Rubiales, *ibid.*, 27, 171 (1995).
- F. Palacios, D. Aparicio and J. M. de los Santos, *Tetrahedron Lett.*, 34, 3481 (1993); F. Palacios, D. Aparicio and J. M. de los Santos, *Tetrahedron*, 50, 12727 (1994).
- F. Palacios, I. Pérez de Heredia and G. Rubiales, J. Org. Chem., 60, 2384 (1995); F. Palacios, C. Alonso and G. Rubiales, *Tetrahedron*, 51, 3683 (1995).
- F. López-Ortiz, E. Pelaez-Arango, F. Palacios, J. Barluenga, S. García-Granda, B. Tejerina and A. García-Fernández, J. Org. Chem., 59, 1984 (1994); F. Palacios, D. Aparicio and J. García, Synlett, 260 (1994).
- 20. F. Palacios, A. Ochoa de Retana and J. Pagalday, Heterocycles, 40, 543 (1995).

1,3-DIPOLAR CYCLOADDITION OF AZIDOALKYLPHOSPHONATES AND CARBOXYLATES

- 21. R. Huisgen, G. Szeimies and L. Möbius, Chem. Ber., 99, 475 (1966).
- 22. R. Fusco, G. Bianchetti and D. Pocar, Gazz. Chem. Ital., 91, 849 (1961).
- 23. R. Huisgen and F. Palacios, Tetrahedron Lett., 23, 55 (1982).
- 24. Y. Kobayashi, A. Ando, K. Kawada, A. Ohsawa and I. Kumadaki, J. Org. Chem., 45, 2962 (1980); Y. Kobayashi, A. Ando, K. Kawada and I. Kumadaki, *ibid.*, 45, 2968 (1980).
- 25. Thermolysis of bicyclic triazolines formed by 1,3-cycloaddition reaction of arylazides and *trans*cyclooctene underwent also ring contraction from the eight membered ring and a seven carbocyclic ring was isolated.²⁶
- 26. R. Huisgen and F. Palacios, Unpublished results.

(Received April 25, 1995; in revised form July 20, 1995)